# [4+2] Diels-Alder cycloaddition reaction of 2-benylidineamino-4-phenyl-1,3-thiazoles with sulfene and their antifungal activities 

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2-Benzylidineamino-4-phenyl-1,3-thiazole undergoes [4+2] Diels-Alder cycloaddition reaction with sulfene resulting in good yield of a mixture of isomeric 2,6 -diphenyl- $2 H, 4 H$-[1,3]thiazolo[3,2-c][1,3,5]thiadiazine 3,3-dioxides and 3, 7-diphenyl[1,3]thiazolo[3,2-b] [1,2,4]thiadiazine 5,5-dioxides derivatives are reported.

Keywords: Diels-Alder reaction, sulfene, thiadiazine-dioxides, isomers

The 1,2,4 and 1,3,5-thiadiazine and its derivatives are an important class of compounds which possess widespread pharmacological properties such as insecticidal and accaricidal ${ }^{1}$ activity. They are also reported to have antifungal ${ }^{2}$ and antiatherosclerotic activity. ${ }^{3}$ Recently Evain et al. has reported ${ }^{4}$ the first example of thiazolo thiadiazine family. 2-Aminothiazoles is the versatile intermediate for highly functionalised heterocycles of widespread pharmaceutical use. ${ }^{5}$ Our interest in synthesis and biological activities of thiadiazine dioxide molecules led us to synthesis of several isomeric 1,3-thiazolo thiadiazine dioxide derivatives from 2-amino-4-phenyl-1,3-thiazole.
Among all the strategies described in the literature for the construction of heterocycles, the [4+2] Diels-Alder cycloaddition reaction between a diene and a dienophile is one of the most versatile routes. ${ }^{6,7,8}$ It has been observed that [ $4+2$ ] Diels-Alder cycloaddition reaction involving sulfene is very rare. Sulfenes have received little attention as dienophiles. 9 ,10 The suitable use of diene and dienophiles determines the wide range of cycloadducts. Literature reports ${ }^{7}$ shows that $[4+2]$ cycloaddition reaction involving 1,3-diazabutadienes with heterodienophile sulfene gives only one cycloadduct.
In this paper, we have reported the synthesis of several previously unknown isomeric 2,6-diphenyl-2 $H, 4 H-[1,3]$ thiazolo [3,2-c] [1,3,5]thiadiazine 3,3-dioxides (2a-2e) and 3,7-
diphenyl[1,3]thiazolo[3,2-b][1,2,4]thiadiazine 5,5-dioxides (3a-3e) in high yield by [4+2] cycloaddition reaction from 2-benzylidineamino-4-phenyl-1,3-thiazole and methanesulfonyl chloride in presence of triethylamine and subsequent separation of the isomers by column chromatography. The column chromatography (silicagel, 9:1 petroleum ether: ethyl acetate) of the crude product afforded two fractions 2 and 3. The first fraction 2 (major product, $70 \% \mathbf{2 a}$, TLC, $\mathrm{R}_{\mathrm{f}}=0.61$ ) and the second fraction $\mathbf{3}$ (minor product, $20 \%, \mathbf{3 a}$, $\mathrm{R}_{\mathrm{f}}=0.12$ ). (Scheme 1 and Table 1).

The structure of the compounds has been assigned $\mathbf{2}$ and 3 on the basis of analytical and spectroscopic (IR, ${ }^{1} \mathrm{H}$ NMR, MS) data. The IR spectra of 2a showed characteristic $\mathrm{SO}_{2}$ stretching frequency around 1167 and 1370 and for $\mathbf{3 a}$ around 1178 and $1330 \mathrm{~cm}^{-1}$ respectively.
The ${ }^{1} \mathrm{H}$ NMR spectra (in $\mathrm{CDCl}_{3}, \delta \mathrm{ppm}$ ) of 2 showed methine proton absorption at $\delta 3.55-4.00$ and for methylene proton $\left(-\mathrm{CH}_{2}-\right)$ at around $\delta 3.19-3.40$, for aromatic proton $\delta 6.46-7.24(\mathrm{~m}, \mathrm{Ar}-\mathrm{H})$ and for $\mathrm{C}-2$ proton singlet at $\delta 8.10 \mathrm{ppm}$ appeared.The ${ }^{1} \mathrm{H}$ NMR spectra for 3 showed methylene proton absorption at $\delta 3.64$ as doublet, methine proton as multiplet at $\delta 4.90 \mathrm{ppm}$, aromatic proton appeared at $\delta 6.40-7.35 \mathrm{ppm}$ and C-2 proton appeared as singlet at $\delta 8.20 \mathrm{ppm}$. It has been found that the product $\mathbf{3}$ is cis as $J$ of C-6 and C-7 is 5 Hz . Mass spectra of 2a showed molecular


Scheme 1
Table 1 Physical data of compound $\mathbf{2}$ and $\mathbf{3}$

| Product | R | Yield/\% | m.p. ${ }^{\circ} \mathrm{C}$ | $\mathrm{R}_{\mathrm{f}}{ }^{\text {a }}$ | $m / z$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 2a | H | 70 | 135 | 0.61 | 342(100\%) |
| 3a | H | 20 | 128 | 0.12 | 342(33\%) |
| 2b | 4-OMe | 63 | 112 | 0.74 | 372(100\%) |
| 3b | 4-OMe | 17 | 127 | 0.22 | 372(25\%) |
| 2c | $4-\mathrm{NO}_{2}$ | 65 | 98 | 0.59 | 387(100\%) |
| 3c | $4-\mathrm{NO}_{2}$ | 19 | 84 | 0.19 | 387(15\%) |
| 2d | $4-\mathrm{NMe}_{2}$ | 63 | 108 | 0.63 | 385(100\%) |
| 3d | $4-\mathrm{NMe}_{2}$ | 19 | 122 | 0.11 | 385(12\%) |
| 2e | $4-\mathrm{OH}$ | 53 | 103 | 0.53 | 358(100\%) |
| 3 e | $4-\mathrm{OH}$ | 29 | 98 | 0.35 | 358(20\%) |
| 2 f | $4-\mathrm{Cl}$ | 68 | 92 | 0.68 | 376(100\%) |
| 3 f | $4-\mathrm{Cl}$ | 18 | 105 | 0.21 | 376(33\%) |

asolvent system for TLC: Petroleum ether, ethyl acetate (7:3).

[^0]Table 2 Analytical and spectroscopic data for compounds 2 and 3

| Compd | Molecular formula | Found (Required)/\% |  |  | $\mathrm{IR}\left(\mathrm{KBr}, \mathrm{\cup} \mathrm{~cm}^{-1}\right)$ | ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, \delta=0 \mathrm{ppm}\right)$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | C | H | N |  |  |
| 2a | $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2}$ | 59.52 | 4.16 | 8.24 | 1167,1370 | $3.20\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2}\right) ; 3.65\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}\right) ;$ |
|  |  | (59.63 | 4.12 | 8.18) |  | $6.46-7.20\left(\mathrm{~m}, 11 \mathrm{H}, \mathrm{Ar}-\mathrm{H}\right.$ and $\left.\mathrm{C}_{2}-\mathrm{H}\right)$ |
| 3a |  | 59.24 | 3.94 | 7.95 | 1178,1330 | 3.64(d, $2 \mathrm{H},-\mathrm{CH}_{2}$ ); 4.90(dd, $1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}$ ); |
|  |  | (59.63 | 4.12 | 8.18) |  | $6.40-7.32\left(\mathrm{~m}, 11 \mathrm{H}, \mathrm{Ar}-\mathrm{H}\right.$ andC $\left.\mathrm{C}_{2}-\mathrm{H}\right)$ |
| 2b | $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}_{2}$ | 58.14 | 4.06 | 7.42 | 1165,1372 | 3.40 (s,2H, $-\mathrm{CH}_{2}$ ); 3.78(s,3H,OMe); |
|  |  | (58.04 | 4.33 | 7.52) |  | $3.84\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}\right) ; 6.20-7.60(\mathrm{~m}, 10 \mathrm{H},$ $\left.\mathrm{Ar}-\mathrm{H} \text { andC } \mathrm{C}_{2}-\mathrm{H}\right)$ |
| 3b |  | 57.80 | 4.54 | 7.28 | 1174,1334 | 3.70(d, $2 \mathrm{H},-\mathrm{CH}_{2}$ ); 3.82(s, $3 \mathrm{H}, \mathrm{OMe}$ ); |
|  |  | (58.04 | 4.33 | 7.52) |  | 4.92(dd, $1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}$ ); 6.35- |
|  |  |  |  |  |  | $7.74(\mathrm{~m}, 10 \mathrm{H}$ |
|  |  |  |  |  |  |  |
| 2c | $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}_{4} \mathrm{~S}_{2}$ | 52.86 | 3.21 | 10.58 | 1165,1371 | 3.40(s,2H,- $\mathrm{CH}_{2}$ ); 4.00(s,1H, $\mathrm{C}_{7}-\mathrm{H}$ ); |
|  |  | (52.70 | 3.38 | 10.85) |  | 6.69-7.95(m,10H, $\mathrm{Ar}-\mathrm{H}$ andC ${ }_{2}-\mathrm{H}$ ) |
| 3c |  | 52.48 | 3.54 | 11.18 | 1170,1335 | 3.72 (d, $2 \mathrm{H},-\mathrm{CH}_{2}$ ); 4.92(dd, $1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}$ ); |
|  |  | (52.70 | 3.38 | 10.85) |  | 6.62-7.96(m, $10 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ andC $\left.\mathrm{C}_{2}-\mathrm{H}\right)$ |
| 2d | $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{~S}_{2}$ | 59.12 | 4.98 | 10.72 | 1168,1374 | 2.85(s, $6 \mathrm{H}, \mathrm{NMe}_{2}$ ); 3.22(s,2H,- $\mathrm{CH}_{2}$ ); |
|  |  | (59.20 | 4.97 | 10.90) |  | $\begin{aligned} & 3.55(\mathrm{~s}, 1 \mathrm{H},-\mathrm{CH}-) ; 6.50-7.40(\mathrm{~m}, 10 \mathrm{H} \text {, } \\ & \left.\mathrm{Ar}-\mathrm{H} \text { and } \mathrm{C}_{2}-\mathrm{H}\right) \end{aligned}$ |
| 3d |  | 58.96 | 5.04 | 11.26 | 1178,1331 | 2.90 (s, $6 \mathrm{H}, \mathrm{NMe}_{2}$ ); 3.92(d, $2 \mathrm{H},-\mathrm{CH}_{2}$ ); |
|  |  | (59.20 | 4.97 | 10.90) |  | 4.90 (dd, $1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}$ ); 6.62- |
|  |  |  |  |  |  | 7.76(m,10H, |
| 2e | $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}_{2}$ | 56.68 | 3.76 | 7.94 | 1167,1375 | $\left.\mathrm{Ar}-\mathrm{H}, \mathrm{C}_{2}-\mathrm{H}\right)$ $3.19\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2}\right) ; 3.70\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}\right)$; |
|  |  | (56.97 | 3.94 | 7.82) |  | $\begin{aligned} & 5.62(\mathrm{~s}, 1 \mathrm{H},-\mathrm{OH}) ; 6.30-7.30(\mathrm{~m}, 10 \mathrm{H}, \\ & \left.\mathrm{Ar}-\mathrm{H} \text { and } \mathrm{a}_{2}-\mathrm{H}\right) \end{aligned}$ |
| 3 e |  | 57.24 | 4.26 | 8.04 | 1174,1332 | $3.82\left(\mathrm{~d}, 2 \mathrm{H},-\mathrm{CH}_{2}\right) ; 4.92\left(\mathrm{dd}, 1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}\right)$; |
|  |  |  |  |  |  | 5.74(s,1H,-OH); |
|  |  | (56.97 | 3.94 | 7.82) |  | 6.25-7.42(m,10H,Ar-H, $\mathrm{C}_{2}-\mathrm{H}$ ) |
| 2 f | $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2} \mathrm{Cl}$ | $54.00$ | $3.56$ | 7.29 | 1165,1372 | $3.32\left(\mathrm{~s}, 2 \mathrm{H},-\mathrm{CH}_{2}\right) ; 3.72\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}\right) ;$ |
|  |  | (54.17 | 3.48 | 7.43) |  | $6.52-7.48(\mathrm{~m}, 10 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ andC 2 - H ) |
| 3 f |  | 54.34 | 3.62 | 7.70 | 1178,1330 | 3.84(d, $2 \mathrm{H},-\mathrm{CH}_{2}$ ); 4.92(dd, $1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}$ ); |
|  |  | (54.17 | 3.48 | 7.43) |  | 6.62-7.60(m, $10 \mathrm{H}, \mathrm{Ar}-\mathrm{H}$ and $\mathrm{C}_{2}-\mathrm{H}$ ) |

ion peak at $m / z 342\left(\mathrm{M}^{+}, 100 \%\right)$ and 3a showed molecular ion peak at $m / z 342\left(\mathrm{M}^{+}, 33 \%\right)$.

## Experimental

M.p.s were determined on a Buchi apparatus. Mass spectra were recoded with a LC-MS, Bruker (Model Esquire-3000) mass spectrometer. IR spectra were recorded on a Perkin-Elmer system 2000 FTIR spectrometer and ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a varian T-60/90 MHz spectrometer.

Preparation of 2,6-diphenyl-2H,4H-[1,3]thiazolo[3,2-c][1,3,5] thiadiazine 3,3-dioxide (2a) and 3,7-diphenyl[1,3]thiazolo[3,2-b] [1,2,4]thiadiazine 5,5-dioxide (3a): In a typical experiment, to a solution of 2-benzylidineamino-4-phenyl-1,3-thiazole ( 0.528 g , 0.002 mol ), in dry 1,4-dioxane ( 20 ml ) was added triethylamine $(0.56 \mathrm{ml}, 0.004 \mathrm{~mol})$ and to this solution was added methanesulfonyl chloride ( $0.23 \mathrm{ml}, 0.003 \mathrm{~mol}$ ) in 1,4-dioxane ( 15 ml ) drop by drop $0-5^{\circ} \mathrm{C}$ under stirring during 30 min . The stirring was continued for a period of $3-4 \mathrm{hr}$. at the same temperature. The reaction mixture was then poured into ice ( 100 gm ) and extracted with chloroform $(2 \times 40 \mathrm{ml})$ and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Removal of the solvent under reduced pressure gave the crude product in $80-90 \%$ yield. The crude products were separated by column chromatography (silica gel, 9:1, petroleum ether: ethyl acetate) afforded 2a and 3a, the product, 2a, yield $70 \%$, m.p. $135^{\circ} \mathrm{C}$ and the product $\mathbf{3 a}$, yield $20 \%$, m.p. $128^{\circ} \mathrm{C}$. For 2a, IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ) 1370, $1167\left(\mathrm{SO}_{2}\right)$; ${ }^{1} \mathrm{H}$ NMR ( $\delta \mathrm{ppm}, \mathrm{CDCl}_{3}$ ) 3.20(s,2H,-CH2), 3.65(s,1H,-C7-H), 6.46$7.20\left(\mathrm{~m}, 11 \mathrm{H}, \mathrm{Ar}-\mathrm{H}\right.$ and $\left.\mathrm{C}_{2}-\mathrm{H}\right)$; mass $\mathrm{m} / \mathrm{z} 342\left(\mathrm{M}^{+}, 100 \%\right)$. For 3a, IR $\left(\mathrm{KBr}, \cup \mathrm{cm}^{-1}\right) 1330,1178\left(\mathrm{SO}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(\delta \mathrm{ppm}, \mathrm{CDCl}_{3}\right) 3.64(\mathrm{~d}, 2 \mathrm{H},-$ $\left.\mathrm{CH}_{2}\right), 4.90\left(\mathrm{dd}, 1 \mathrm{H}, \mathrm{C}_{7}-\mathrm{H}\right), 6.40-7.32\left(\mathrm{~m}, 11 \mathrm{H}, \mathrm{Ar}-\mathrm{H}\right.$ and $\left.\mathrm{C}_{2}-\mathrm{H}\right)$; mass $m / z 342\left(\mathrm{M}^{+}, 33 \%\right)$.

## Biological activity screening

Due to the diverse and interesting biological activities of $1,2,4$ and 1,3,5-thiadiazine ring system, we screened these compounds for their antifungal activities against Rhizoctonia solani and Drechslera oryzae, two important fungal pathogens causing diseases on rice crop. Their antifungal activities were evaluated according to the inhibition zone technique. ${ }^{12}$ It has been observed that the
thiazolothiadiazine dioxide compounds have excellent antifungal activities. Comparatively significant activity was observed in compounds bearing substituent $\mathrm{NMe}_{2}$ (compound no. 2d and 3d) at a concentration of $1 \mathrm{mg} / \mathrm{ml}$. The control (without treatment) sets of experiments with both the fungal pathogens exhibited no inhibition of growth. However, the standard fungicide (carbendazim) showed 98.56 and $98.26 \%$ inhibition of Rhizoctonia solani and Drechslera oryzae respectively at the same concentration of $1 \mathrm{mg} / \mathrm{ml}$.

In conclusion, we have synthesised an important class of isomeric thiazolo-1,2,4 and 1,3,5-thiadiazine derivatives from easily prepared 2-amino-4-phenyl-1,3-thiazole.

We thank the Director, RRL, Jorhat, for providing the necessary facilities to carry out this research work, also Dr J.C.S. Kataky Scientist-F, of this laboratory, for his useful suggestion during this research work.

Received 29 December 2006; accepted 27 February 2007
Paper 06/4374 doi: 10.3184/030823407X191877

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